LEVELIT NBSIR-78-1473 A045 095 A 056 **Optical Materials** <u>Characterization</u> AD Albert Feldman, Deane Horowitz, Roy M. Waxler and Marilyn J Dodge Technical Next. 1 a. 71-31 8.78 Ceramics, Glass and Solid State Science Division Center for Materials Science National Bureau of Standards Washington, D.C. 20234 May 1978 Period Covered: August 1, 1977 to January 31, 1978 ARPA Order 12620 DISTRUMITION STATEMENT A Approved for public relaced Matribution Unlimited

Prepared for

Advanced Research Project Agency

Arlington, Virginia 22209

41475\$ **8 06 22 098** 

NBSIR 78-1473

# OPTICAL MATERIALS CHARACTERIZATION

Albert Feldman, Deane Horowitz, Roy M. Waxler and Marilyn J. Dodga

Ceramics, Glass and Solid State Science Division Center for Materials Science National Bureau of Standards VVashington, D.C. 20234

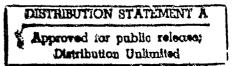
May 1978

Period Covered: August 1, 1977 to January 31, 1978

ARPA Order No: 2620



Prapared for Advanced Research Project Agency Arlington, Virginia 22209







U.S. DEPARTMENT OF COMMERCE, Juanita M. Kreps, Secretary

Dr. Sidney Harman, Under Secretary

Jordan J. Baruch, Assistant Secretary for Science and Technology NATIONAL BUREAU OF STANDARDS, Emest Ambier, Director

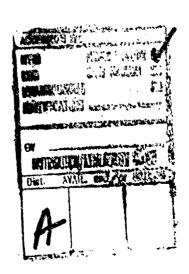
#### OPTICAL MATERIALS CHARACTERIZATION

Albert Feldman, Deane Horowitz, Roy M. Waxler, and Marilyn J. Dodge

Ceramics, Glass & Solid State Science Division Center for Materials Science

ARPA Order No	•	•	•	•	•	•	•	:	•	•	2620
Program Code Number					•	•	•	•		•	4D10
Effective Date of Contract									•		January 1, 1974
Contract Expiration Date .		•			•		•				September 30, 1978
Principal Investigator	•		•	•		•	•	•			Albert Feldman (301) 921-2840

The views and conclusions contained in this document are those of the authors and should not be interpreted as necessarily representing the official policies, either expressed or implied, of the Advanced Research Projects Agency or the U.S. Government. The results in this report are preliminary in nature and are subject to change.



# Table of Contents

1.	Tech	nical Report Summary	1
	1.1	Technical Problem	1
	1.2	General Methodology	1
	1.3	Technical Results	2
	1.4	Department of Defense Implications	2
	1.5	Implications for Further Research	2
2.	Tech	nical Report	3
	2.1	Thermo-optic and Linear Thermal Expansion Coefficients	3
	2.2	Piezo-optic Constants	4
3.	Ackn	owledgement	7

#### OPTICAL MATERIALS CHARACTERIZATION

MICROMETTIS
Abstract

The piezo-optic constants of CaFt, BaFt, and SrFt have been measured at 0.6328 (m) and 1.15 (m). The temperature dependence of the refractive indices of CdFt, MgFt, and NaCl have been measured at several wavelengths in the infrared by the method of Fizeau interferometry. The linear thermal expansion coefficients of NaCl and CdFt as a function of temperature have also been measured.

#### OPTICAL MATERIALS CHARACTERIZATION

#### 1. Technical Report Summary

### 1.1 Technical Problem

Windows subjected to high-average-power laser radiation will undergo optical and mechanical distortion due to absorptive heating. If the distortion becomes sufficiently severe, the windows become unusable. Theoretical calculations of optical distortion in laser windows depend on the following material parameters; absorption coefficient, refractive index, change of index with temperature, thermal expansion coefficient, stress-optical constants, elastic compliances, specific heat, thermal conductivity and density. Our program has been established to measure refractive indices, changes of index with temperature, stress-optical constants, elastic compliances, and thermal expansion coefficients of candidate laser window materials.

## 1.2 General Methodology

Laboratory experiments are conducted for measuring refractive indices, changes of index with temperature, stress-optical constants, elastic compliances, and thermal expansion coefficients.

The refractive indices of prismatic specimens are measured on precision spectrometers by using the method of minimum deviation. Two spectrometers are used. One instrument, which uses glass optics, is used for measuring refractive indices in the visible with an accuracy of several parts in 10°. The other instrument, which uses mirror optics, is used for measuring refractive indices in the ultraviolet and the infrared to an accuracy of several parts in 10°. Using both spectrometers we can measure refractive indices with the spectral region 0.2 µm to 50 µm.

We measure the coefficient of linear thermal expansion, a, by a method of Fizeau interferometry. The interferometer consists of a specially prepared specimen which separates two flat plates. Interference fringes are observed due to reflections from the plate surfaces in contact with the specimen. We obtain a by measuring the shift of these interference fringes as a function of temperature. We can measure a from -180 °C to 800 °C.

The change of refractive index with temperature, dn/dT, is measured by two methods. In the first method, we measure the refractive index with the precision spectrometers at two temperatures, 20 °C and 30 °C, by varying the temperature of the laboratory. This provides us with a measure of dn/dT at room temperature. The second method may be used for measuring dn/dT from 180 °C to 800 °C. We obtain dn/dT from a knowledge of the expansion coefficient and by measuring the shift of Fizeau

fringes in a heated specimen as a function of temperature. The Fizeau fringes are due to interferences between reflections from the front and back surfaces of the specimens.

We measure piezo-optic coefficients and elastic compliances using a combination of Twyman-Green and Fizeau interferometers. From the shift of fringes in specimens subjected to uniaxial or hydrostatic compression, we obtain the necessary data for determining all the stress-optical constants and elastic compliances.

In materials with small piezo-optic constants or in materials that cannot withstand large stresses, we use interferometers designed to measure fractional fringe shifts. At 10.6  $\mu$ m we use a modified Twyman-Green interferometer which has a sensitivity of 0.01 $\lambda$ . At 632.8 nm, we use a modified Dyson interferometer which has a sensitivity of 0.002 $\lambda$ . When using these interferometers to measure piezo-optic constants we must know the elastic constants of the material under test.

### 1.3 Technical Results

The temperature dependences of the thermo-optic coefficients of  $CdF_2$ ,  $MgF_2$  and NaCl have been measured over the temperature range -180 °C to 200 °C at discrete wavelengths in the infrared by the method of Fizeau interferometry. The linear thermal expansion coefficients of  $CdF_2$  and NaCl were also measured over the same temperature range. (Section 2.1)

The piezo-optic constants  $q_{11}$ ,  $q_{12}$ , and  $q_{44}$ , of CaF<sub>2</sub>, BaF<sub>2</sub> and SrF<sub>2</sub> have been measured at 0.6328  $\mu m$  and 1.15  $\mu m$ . (Section 2.2)

#### 1.4 Department of Defense Implications

The Department of Defense is currently constructing high-power laser systems. Criteria are needed for determining the suitability of different materials for use as windows in these systems. The measurements we are performing provide data that laser system designers can use for determining the optical performance of candidate window materials.

#### 1.5 Implications for Further Research

We plan to measure the refractive indices of SrF. and MgF, from the ultraviolet into the infrared. Measurements of the thermo-optic coefficients of LiF, NaF, MgF, CdF, NaCl, Al<sub>2</sub>O<sub>3</sub>, CaF, BaF, SrF, KCl, and KBr are planned for the wavelengths 458 nm and 350 nm. Piezo-optic coefficient measurements are planned for SiO<sub>2</sub>, CaF, and Al<sub>2</sub>O<sub>3</sub> at 350 nm. All this work cannot be done by September 30, 1978, however, we will do as much as possible under the constraints of the funding available.

#### 2. Technical Report

# 2.1 Thermo-optic and Linear Thermal Expansion Coefficients

In addition to the previously reported [1] dn/dT data on  $CaF_2$ ,  $BaF_2$ , KBr(RAP), KC1(RAP), LiF, NaF,  $SrF_2$ , ZnS(CVD), and ZnSe(CVD), dn/dT was measured on single crystals of  $CdF_2$ ,  $MgF_2$ , and NaC1.

Figure 1 shows a plot of dn/dT as a function of temperature for  $CdF_2$  at the two helium-neon laser wavelengths, 0.6328  $\mu m$  and 3.39  $\mu m$ , for the temperature range, -180 °C to 200 °C. The solid line curve represents a least squares, third order polynomial fit to the .6328  $\mu m$  dn/dT data. Table 1 presents the results of this fit and a similar third order polynomial fit at 3.39  $\mu m$  in a tabulated form with 20 °C temperature intervals. The errors in the table are the standard deviation of the experimental data to the least squares fit.

dn/dT as a function of temperature for NaCl is shown in Figure 2 for the three wavelengths, 0.6328  $\mu m$ , 1.15  $\mu m$ , and 3.39  $\mu m$ . The results of a least squares third order polynomial fits for each wavelength are presented in Table 2.

MgF<sub>2</sub>, which is an anisotropic crystal, was measured at 0.6328 µm and 5.39 µm with the electric field parallel to the c-axis to get dn /dT, and with the electric field perpendicular to the c-axis to get dn /dT. The birefringence as a function of temperature, d(n-n)/dT, was also measured at 0.6328 µm. The birefringence as a function of temperature was too small to be measured at 3.39 µm. The upper set of points in Figure 3 shows dn /dT as a function of temperature at 0.6328 µm and 3.39 µm, the middle set of points shows dn /dT as a function of temperature at 0.6328 µm and 3.39 µm, and the lower set of points shows the experimentally measured birefringence, d(n-n)/dT, at 0.6328 µm. The straight lines represent a linear least squares fit to the 0.6328 µm data in which there was the constraint that the difference in the dn /dT and dn /dT fits equals the birefringence fit, d(n-n)/dT. Table 3 shows the tabulated results for MgF<sub>2</sub>.

The refractive indices, the specimen thicknesses, and references to the thermal expansion coefficient, all of which were used in the computation of the results for each of the materials, are given in Table 4.

The linear thermal expansion coefficients of NaCl and CdF, were also measured. Figure 4 gives the thermal expansion of NaCl in which the triangles are the experimental results and the circles are the AIP handbook values. The dashed line represents the fit to our data, which was used to calculate dn/dT for NaCl. Figure 5 gives the thermal expansion of CdF, as a function of temperature. Only one reference to the thermal expansion of CdF, was found.

#### 2.2 Piezo-optic Constants

The piezo-optic constants have been treated amply in the literature [2] so that it is not necessary to describe them here. The rare-earth fluorides are cubic belonging to the crystal class m3m and have three piezo-optic coefficients,  $q_{11}$ ,  $q_{12}$ , and  $q_{44}$ . These coefficients have been evaluated at two wavelengths, 0.6328 µm and 1.15 µm by measuring the changes in optical path length induced by compressive loading on specimens in the shape of rectangular prisms. Helium-neon laser sources were used at both wavelengths, and the optical path change was measured interferometrically by noting the shift in interference fringes. The fringes were detected by a silicon matrix vidicon camera and observed on a television monitor. Determinations were made for CaF<sub>2</sub>, SrF<sub>2</sub> and BaF<sub>2</sub>.

The specimens, which were obtained commercially, had been precision ground to the approximate dimensions,  $38\text{mm} \times 13\text{mm} \times 13\text{mm}$ . The method of mounting and loading the specimens has been described earlier [3]. Two specimens of each material were fabricated. In the first specimen, the longest dimension was parallel to the <001> crystallographic direction and the light was propagated parallel to the <010> direction. In the second specimen the longest dimension was parallel to the <111> direction and the light was propagated along the <\bar{1}0> direction.

Two opposite long faces of each prism had been polished sufficiently flat and parallel so that about six localized, Fizeau-type interference fringes could be observed across the face when illuminated with collimated monochromatic light. At the wavelength, 0.6328  $\mu m$ ,  $q_1$  and  $q_2$  were determined by measuring the shift in these Fizeau fringes with load on the <001> specimens. The coefficient  $q_{d,1}$  was obtained from stress birefringence measurements on a <111> specimen. The optical set-up and equations relating the changes in refractive index with stress have been presented elsewhere [3-6].

At 1.15 µm, the use of a <001> prism for the determination of  $q_{11}$  and  $q_{12}$  was found to be inadequate because of the small shift in interference fringes with load; instead, the <111> prism was used. Measurements were made of the shift in interference fringes for light polarized both vertically and horizontally.  $q_{11}$  and  $q_{12}$  were then evaluated by solving simultaneously the two equations

$$\Delta n_1 = n_0^{-3} (q_{11} + 2q_{12} + 2q_{44}) \frac{p}{6}$$
 (1)

and

$$^{\Delta n_2} = ^{n_0 3} (q_{11} + 2q_{12} - q_{44}) \frac{p}{6}$$
 (2)

where  $\Delta n_1$  and  $\Delta n_2$  are respectively, the refractive index changes for light polarized vertically and horizontally,  $n_1$  is the initial refractive index, and P is the applied stress. To determine  $q_{11}$  and  $q_{12}$  from the above equations, it is necessary to know  $q_{44}$ , and this value was found by measuring the stress induced birefringence in the <111> specimen.

The results of the study are presented in Table 5. The data indicate that there is little dispersion between the determinations at 0.6328  $\mu m$  and 1.15  $\mu m$ . For comparison, data from the literature [7] taken at 0.6328  $\mu m$  are also presented. Except for  $q_{11}$  and  $q_{12}$  in CaF $_2$ , the disagreement of our data with the data in the literature is significant. We suspect that many of the deviations observed may be due to erroneous values of the elastic constants in the literature. These constants are used in the analysis of the interferometric data in order to obtain the piezo-optic coefficients. They are also used in the conversion of elasto-optic coefficients to piezo-optic coefficients.

An example of the difficulty with the elastic compliances arose in the measurement of the piezo-optic constants of SrF. It was found that the coefficients  $\mathbf{q}_{11}$  and  $\mathbf{q}_{12}$  obtained on the <001> Specimen differed from the values obtained on the <111> specimen. The discrepancy was resolved by the performance of measurements on a Twyman-Green interferometer [8] in addition to the Fizeau interferometer measurements. Both sets of measurements permitted us to calculate elastic compliance components which differed from values in the literature. These values will be presented in a future report.

#### References

- A. Feldman, D. Horowitz, R. M. Waxler, M. J. Dodge, and W. K. Gladden, Optical Materials Characterization, National Bureau of Standards International Report, NBSIR 77-1304 (August, 1977).
- 2. J. F. Nye, <u>Physical Properties of Crystals</u> (Oxford University Press, London, 1957), pp. 243-254.
- 3. A. Feldman and W. J. McKear, Rev. Sci. Instrum., 46, 1588 (1975).
- 4. A. Feldman, R. M. Waxler, and D. Horowitz, Optical Properties of Highly Transparent Solids, Ed. by S. S. Mitra and B. Bendow (Plenum Publishing Corp., New York, 1975), pp. 517-525.
- 5. R. M. Waxler and E. N. Farabaugh, J. Res., NBS 74A, 215 (1970).
- 6. A. Feldman, Electro-optical Systems Design, 8, 36 (1976).
- S. K. Dickinson, <u>Infrared Laser Window Materials Property Data</u> for ZnSe, KCl, NaCl, CaF<sub>2</sub>, SrF<sub>2</sub>, BaF<sub>2</sub>, (AFCRL-TR-75-0318 Physical Sciences Research Papers, No. 635, Solid State Sciences Laboratory, Projects 5620,3326, Air Force Cambridge Research Laboratories, L. G. Hanscom Field, Bedford, Massachusetts 01730, June 6, 1975), pp. 147-194.
- N. Born and E. Wolf, <u>Principles of Optics</u> (Pergamon Press, 1970), p. 303.

Table 1. dn/dT of  $CdF_2$   $(10^{-5}K^{-1})$ 

<b>.</b>	Wavelength (µm)				
Temperature (°C)	0.6328 <sup>a</sup>	3.39 <sup>b</sup>			
-180	- 0.56	- 0.53			
-160	- 0.64	- 0.64			
-140	- 0.72	- 0.73			
-120	- 0.78	- 0.81			
-100	- 0.84	- 0.87			
- 80	- 0.89	- 0.93			
- 60	- 0.93	- 0.98			
- 40	- 0.97	- 1.02			
- 30	- 1.01	- 1.05			
0	- 1.04	- 1.08			
20	- 1.07	- 1.11			
40	- 1.10	- 1.14			
60	- 1.13	- 1.17			
80	- 1.16	- 1.20			
100	- 1.19	- 1.23			
120	- 1.23	- 1.27			
140	- 1.27	- 1.31			
160	- 1.31	- 1.36			
180	- 1.37	- 1.42			
200	- 1.43	- 1.49			

<sup>&</sup>lt;sup>a</sup>Standard deviation from a third degree polynomial fit is 0.02

bStandard deviation from a third degree polynomial fit is 0.04

Table 2. dn/dT of NaC1  $(10^{-5}K^{-1})$ 

Town and home		Vavelength (µm)	· · · · · · · · · · · · · · · · · · ·
Temperature (°C)	.6328 <sup>a</sup>	1.15 <sup>a</sup>	3.39 <sup>a</sup>
-180	-2.16	-2.22	-2.24
-160	-2.40	-2.48	-2.49
-140	-2.61	-2.70	-2.70
-120	-2.79	-2.89	-2.89
-1.00	-2.96	-3.06	-3.05
- 80	-3.09	-3.20	-3.19
- 60	-3.21	-3.32	-3.31
- 40	-3.32	-3.42	-3.41
- 20	-3.40	-3.51	-3.49
0	-3.48	-3.58	-3.57
20	-3.54	-3.64	-3.63
40	-3.50	-3.70	-3.68
60	-3.65	-3.74	-3.73
30	-3,69	-3.79	-3.78
100	-3.74	-3.83	-3.83
120	-3.78	-3.88	-3.88
140	-3.83	-3.93	-3.94
160	-3.88	-3.99	-4.01
180	-3.94	-4.06	-4.09
200	-4.01	-4.14	-4.18

aStandard deviation from a third degree polynomial fit is 0.04

Table 3. dn/dT of  $MgF_2$   $(10^{-6}K^{-1})$ 

	Wavelength (um)					
emperature (°C)	0.6	328 <sup>a</sup>	3.39 <sup>b</sup>			
	dn <sub>e</sub> /dT	dn <sub>o</sub> /dT	dn <sub>e</sub> /dT	dn <sub>o</sub> /dl		
180	1.65	2.23	1.5	2.0		
-160	1.54	2.12	1.4	2.0		
-140	1.44	2.01	1.3	1.9		
-120	J.33	1.90	1.2	1.8		
-100	1.22	1.79	1.2	1.7		
- 80	1.12	1.68	1.1	1.6		
- 60	1.01	1.57	1.0	1.5		
- 40	0.90	1.46	0.9	1.4		
- 20	0.80	1.35	0.8	1.3		
0	0.69	1.24	0.7	1.2		
20	0.58	1.12	0.6	1.1		
40	0.48	1.01	0.5	1.0		
60	0.37	0.90	0.4	1.0		
80	0.27	0.79	0.3	0.9		
100	0.16	0.68	0.2	0,8		
120	0.05	0.57	0.1	0.7		
140	-0.05	0.46	0	0.6		
160	-0.16	0.35	-0.1	0.5		
180	-0.27	0.33	-0.2	0.4		
200	-0.37	0.13	-0.3	0.3		

aStandard deviation to be determined

<sup>&</sup>lt;sup>b</sup>Standard deviation from a linear fit is 0.2

Table 4. Data used in Computation of dn/dT

Material	Ref	t (mm)	α		
	632.8 nm	1.15 µm .	3.39 µm	(,	
CdF <sub>2</sub>	1.5735 <sup>a</sup>		1.54 <sup>b</sup>	7.33	С
MgF <sub>2</sub> n <sub>e</sub>	1.3887 <sup>d</sup>	1.384 <sup>e</sup>	1.369 <sup>e</sup>	13.40	£
MgF <sub>2</sub> n <sub>o</sub>	1.3770 <sup>d</sup>	1.373 <sup>e</sup>	1.358 <sup>e</sup>	13.40	f
NaC1	1.542 <sup>g</sup>	1.5305 <sup>g</sup>	1.52358	14.08	h

 $<sup>^{\</sup>mathbf{a}}$ B. Krukoska-Fulde, T. Niemyski, J. Crystal Growth  $\underline{\mathbf{1}}$ , 183-6 (1967).

b<sub>Estimated</sub>.

CSee Figure 5.

dA. Duncanson, R. W. H. Stevenson, Proc. Phys. Soc. (London) 72, 1001 (1958).

eH. H. Li, to be published.

fJ. S. Browder, S. S. Ballard, Appl. Optics 16 (12), 3214-7.

gs. S. Ballard, J. S. Browder, J. F. Ebersole, AIP Handbook, Dwight E. Gray ed. (McGraw-Hill Book Co., 1972), pp. 6-12 to 6-57.

hR. K. Kirby, T. A. Hahn, B. D. Rothrock, AIP Handbook, Dwight E. Gray ed. (McGraw Hill Book Co., 1972), pp. 4-119 to 4-142 and Figure 4.

Table 5. Piezo-optic Constants of Three Alkaline Earth Fluorides

	$\lambda = 0.6$	5328 μm	$\lambda = 1.15  \mu \text{m}$		
	NBS <sup>a</sup>	Literature <sup>b</sup>	NBS <sup>a</sup>		
CaF <sub>2</sub>					
q <sub>11</sub>	-0.38±0.03	-0.41	-0.40±0.06		
q <sub>12</sub>	1.08±0.03	1.04	1.09±0.06		
(q <sub>11</sub> -q <sub>12</sub> )	-1.46±0.01	-1.45	-1.49±0.02		
9 <sub>44</sub>	0.71±0.01	0.84	0.72±0.01		
SrF <sub>2</sub>					
9 <sub>11</sub>	-0.64±0.04	-0.58	-0.63±0.05		
q <sub>12</sub>	1.45±0.04	1.77	1.50±0.05		
$(q_{11}^{}-q_{12}^{})$	-2.08±0.01	-2.35	-2.13±0.04		
q <sub>44</sub>	0.60±0.01	6.59	0.62±0.02		
BaF <sub>2</sub>					
q <sub>11</sub>	-C.99±0.03	-0.62	-0.91±0.07		
9 <sub>12</sub>	2.07±0.04	2.31	2.13±0.07		
$(q_{11}^{}-q_{12}^{})$	-3.06±0.01	-2.93	-3.03±0.02		
944	0.95±0 01	1.06	0.95±0.01		

 $<sup>^{\</sup>mathbf{a}} \mathsf{The}$  errors were calculated from the standard deviations of the experimental data.

bReference (6), the data for SrF<sub>2</sub> were calculated from the values of p<sub>ij</sub> and s<sub>ij</sub> given in reference (6).

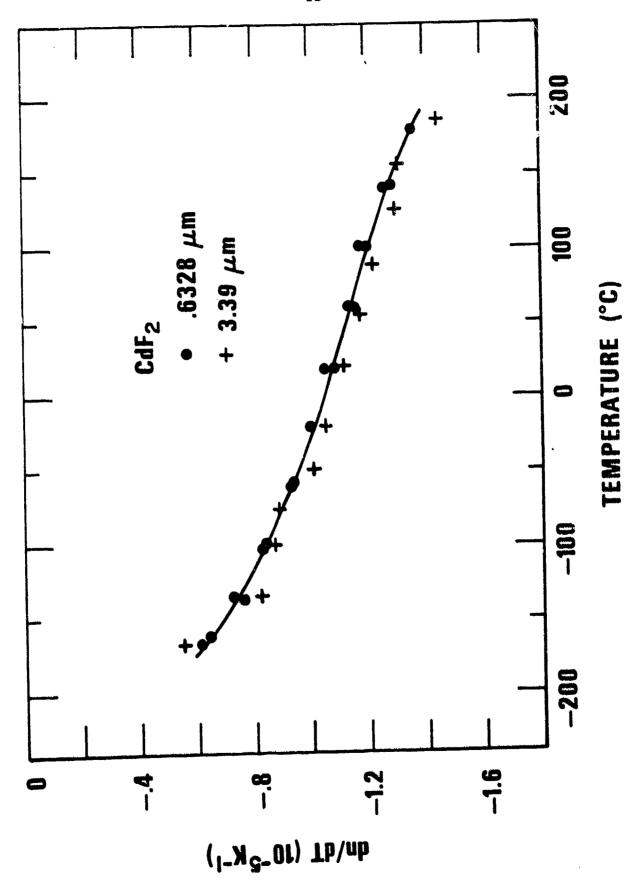


Fig. 1 dn/dT of  $CdF_2$  as a function of temperature.

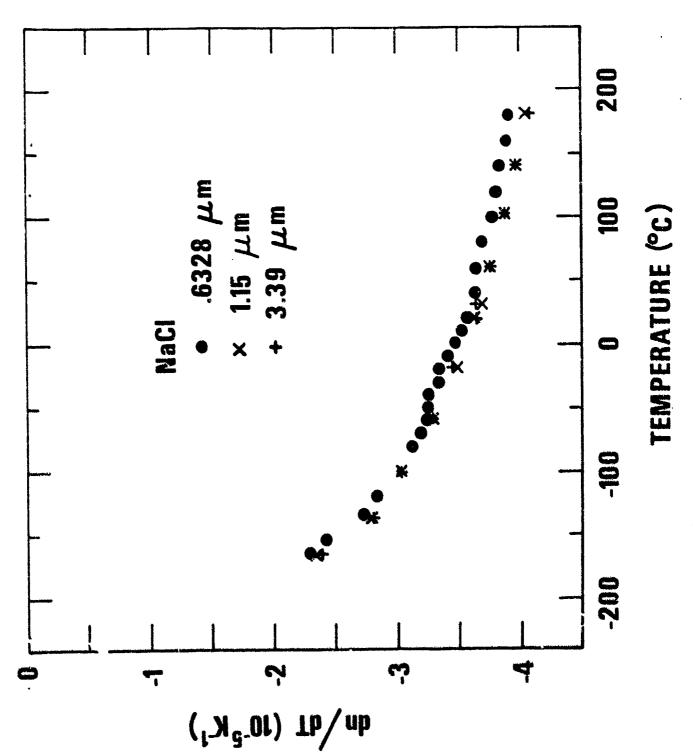


Fig. 2 dn/dT of NaCl as a function of temperature.

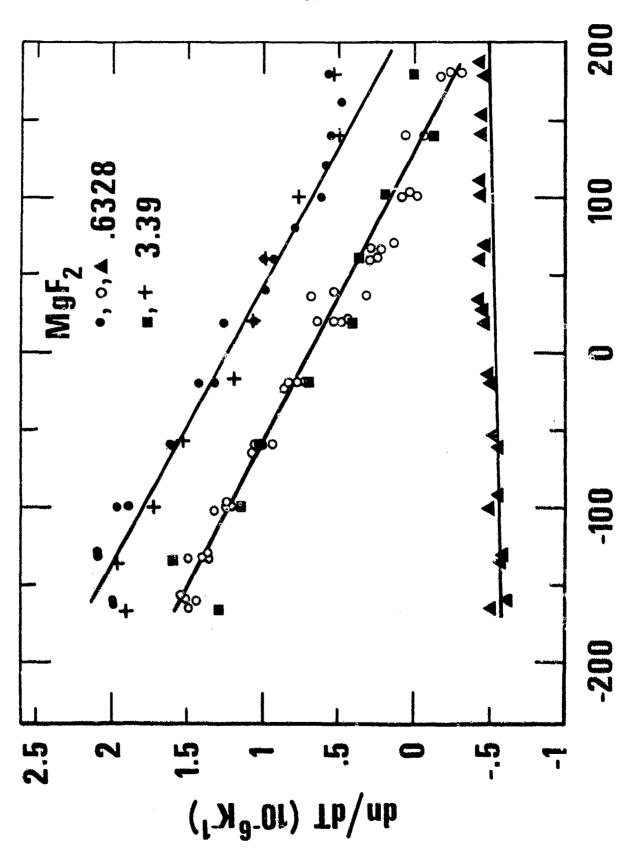


Fig. 3 dn/dT of MgF<sub>2</sub> as a function of temperature. The upper curve is for dn /dT, the middle curve is for dn /dT, and the bottom curve is the birefringence, d(n<sub>e</sub>-n<sub>o</sub>)/dT.

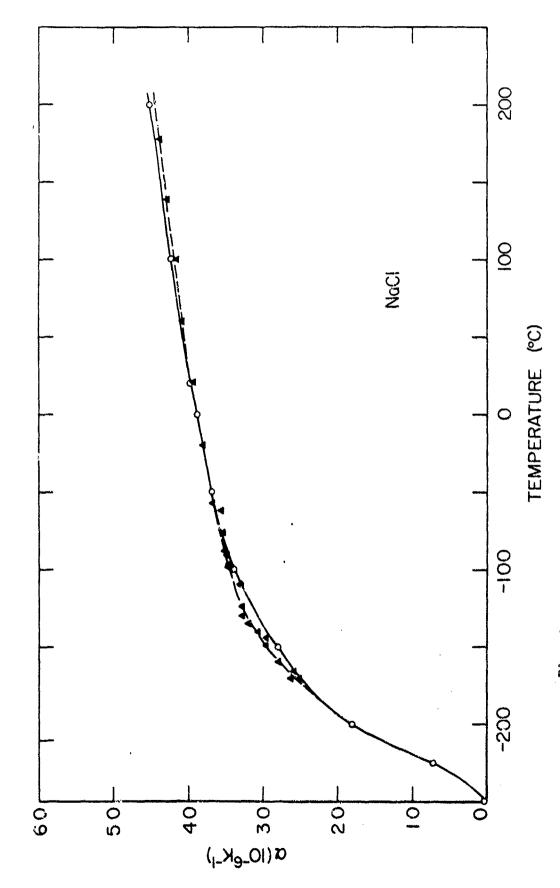
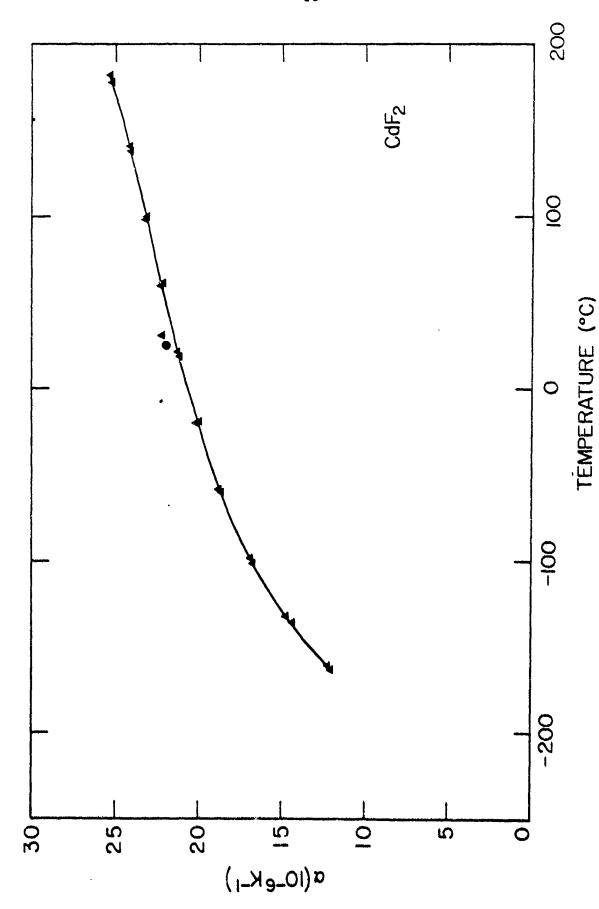


Fig. 4 Thermal expansion coefficient for NaCl as a function of temperature. The triangles are our experimental points and the circles are obtained from the AIP Handbook.



The thermal expansion coefficient for  ${\rm CdF}_2$  as a function temperature. The triangles show our experimental data and the single circle is from S. S. Ballard and J. S. Browder, Appl. Optics 5, 1873 (1966). Fig. 5

# 3. Acknowledgement

We thank Ronald Munro for his assistance with the least squares fit of the  $\ensuremath{\mathrm{MgF}}_2$  data.

U.S. DEPT, OF COMM.  BIBLIOGRAPHIC DATA  SHEET	1. PUBLICATION OR REPORT NO.	2. Gov't Accession No.	3. Recipient	s Accession No.	
4. TITLE AND SUBTITLE	NBSIR-78-1473 (ARPA)		5. Publicatio	- D	
THEE MAD SOUTHEE			J. Publicatio	ii Date	
OPTICAL MATERIALS	CUADA CTEDITA TION				
OPTICAL MATERIALS	CHARACTERIZATION		6. Performing	3 Organization Code	
7. AUTHOR(S) Albert Fe	ldman, Deane Horowitz, Roy M	. Waxler,	8. Performing	Organ, Report No.	
and Maril	yn J. Dodge	· -			
9. PERFORMING ORGANIZAT	ION NAME AND ADDRESS		1	ask/Work Unit No.	
-	BUREAU OF STANDARDS		5650442		
1	IT OF COMMERCE N, D.C. 20234		11. Contract/Grant No.		
WASHING CO	N, D.C. 20234				
12. Sponsoring Organization Na	me and Complete Address (Street, City, St	ate, ZIP)	13. Type of R	eport & Period	
	• • • • • • • • • • • • • • • • • • • •	•	Covered	•	
Advanced	Research Projects Agency				
	, Virginia 22209		14. Sponsorin	g Agency Code	
			[		
15. SUPPLEMENTARY NOTES					
į					
	less factual summary of most significant i	nformation. Il docum <mark>e</mark> n	t includes a si	igniticant	
bibliography or literature su	is say, montton to vasar).				
The niezo-on	tic constants of CaF RaF	and SrE have I	haan mancu	mad at	
0.6328 um and 1.1	tic constants of $CaF_2$ , $BaF_2$ , $5 \mu m$ . The temperature dependent	dence of the re-	Fractive i	ndicae	
of CdF. MgF. an	d NaCl have been measured at	several wavele	noths in t	he	
infrared by the m	ethod of Fizeau interferomet	ry. The linear	thermal e	xpansion	
coefficients of N	aCl and CdF <sub>2</sub> as a function o	f temperature a	re also pr	esented.	
	2	•			
•					
	DISTRIBUTION STATES	ENT A			
	Approved for public of	Angues			
	Distribution Unlimit	-			
				-	
17. KEY WORDS (six to twelve	entries; alphabetical order; capitalize only	the first letter of the l	irst key word	uniess a proper	
name; separated by semicold	ene)				
<b>Thermal</b>		•		MgF <sub>2</sub> ; NaCl.	
inermal coefficien	nt of refractive index; there	20-optic constar	it; linear	thermal	
•	optic constants; photo-elast			SrF_; CdF_;	
18. AVAILABILITY	☐X Unlimited	19. SECURITY		21. NO. OF PAGES	
	B. M. B		- 1787		
For Official Distribution	a. Do Not Release to NTIS	UNCL ASS	IFIED	22	
Onder State State of State	II C Consumers Dalada - Otto -			22 Deice	
Vashington, D.C. 20402	., U.S. Government Printing Office	20. SECURIT		22. Price	
		1			
Order From National Technical Information Service (NTIS)  Springfield, Virginia 22151		1		\$4.00	